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Key indicators

Single-crystal X-ray study T = 293 KMean σ (C–C) = 0.005 Å R factor = 0.026 wR factor = 0.063 Data-to-parameter ratio = 11.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Tetrakis(μ -4-chlorophenoxyacetato- $\kappa^2 O, O'$)dimolybdenum(II)

The crystal structure determination of the title compound, $[Mo_2(C_8H_6ClO_3)_4]$, confirms the existence of a quadruple bond between the Mo atoms, with an Mo–Mo distance of 2.101 (5) Å. One half of the molecule constitutes the asymmetric unit and the carboxylate ligands define a paddlewheel structure, with similar O–C–O angles of 122.9 (3) and 123.7 (3)°. The interplanar angles between the acetate and 4-chlorophenoxy planes in the two independent ligands are 135.9 (8) and 91.3 (8)°, which is indicative of a dramatic difference in the orientation of the ligands.

Comment

Quadruply bonded dimolybdenum complexes have been well established for numerous structures containing carboxylate bridges over the past three decades (Cotton & Walton, 1993). Some specific examples include $[Mo_2(OCOC_7H_7O)_4]$ (Cotton *et al.*, 1983), $[Mo_2(OCOC_4H_{10}N)_4]$ (Belforte *et al.*, 1993), and $[Mo_2(OCOCH_3O)_4]$ (Cotton *et al.*, 2002). We present here the crystal structure of the title dimolybdenum complex, (I).



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Figure 1

A drawing of (I), with displacement ellipsoids shown at the 50% probability level. Unlabelled atoms are related to labelled atoms by the symmetry operator (-x, -y, -z).

The molecule of (I) is arranged around an inversion centre, resulting in one Mo atom and two ligands constituting the asymmetric unit (Fig. 1). The Mo–Mo distance is 2.101 (5) Å, within the range of 2.03–2.25 Å suggested for a quadruply bonded dimolybdenum unit (Cotton & Walton, 1993). Three of the Mo-O distances are similar, at 2.096 (2), 2.0991 (19) and 2.117 (2) Å, with one very different at 2.1529 (19) Å. All other bonds fall within expected values.

The molecule of (I) does not possess any axially bonded solvent molecules, even though it was crystallized under identical conditions to the related compound tetra(4hydroxyphenylacetato)dimolybdenum(II), which contains axially bonded tetrahydrofuran (THF) molecules with an Mo-O(THF) distance of 2.608 (2) Å (Liwporncharoenvong & Luck, 2002). In (I), the molecules are arranged so that there are intermolecular interactions between an Mo atom and one adjacent carboxylate O atom in a neighbouring molecule and their symmetry-generated atoms, forming a parallelogram with Mo–O atom distances of 2.749 (2) Å (Fig. 2), resulting in an infinite chain linkage. These interactions were previously described for $[Mo_2(OCOC_7H_7O)_4]$ (Cotton et al., 1983). As a consequence of this interaction, the carboxylate O atom involved has the longest O-Mo atom bond distance of 2.1529(19) Å, at the upper end of the average of 2.12(2) Å calculated for the 183 dimolybdenum structures bridged by one or more carboxylate ligands listed in the Cambridge Structural Database (CSD, Version 5.26, November 2004 release with February 2005 updates; Allen, 2002) and obtained using the EXCELL program option (Bruno et al., 2002).

The two ligands of (I) contain significantly different torsion angles, with C9-C10-O6-C11 = 74.0 (4)° and C1-C2- $O3-C3 = 175.2 (3)^\circ$. This may be a consequence of the packing and stacking arrangement of the 4-chlorophenoxy groups as the molecule aligns itself to form the crystal (Fig. 2).

Experimental

A solution of K₄Mo₂Cl₈ (300 mg, 0.415 mmol) in distilled water (10 ml) was added to a solution of 4-chlorophenoxyacetic acid (324 mg, 1.74 mmol) in methanol (20 ml). The yellowish product was





A packing diagram for (I) obtained using MERCURY (Bruno et al., 2002), showing the parallelogram formed by weakly bonded interactions (dashed lines) between adjacent molecules. Dashed lines ending in + signs indicate the direction of the infinite chains.

filtered off and vacuum dried. Crystals of (I) were obtained by allowing methanol diffusion into a tetrahydrofuran solution of the product mixture.

Crystal data

$[Mo_2(C_8H_6ClO_3)_4]$	Z = 1
$M_r = 934.19$	$D_x = 1.797 \text{ Mg m}^{-3}$
Triclinic, P1	Mo $K\alpha$ radiation
a = 5.668 (1) Å	Cell parameters from 25
b = 11.549(1) Å	reflections
c = 13.743 (2) Å	$\theta = 10 - 15^{\circ}$
$\alpha = 91.95 \ (1)^{\circ}$	$\mu = 1.10 \text{ mm}^{-1}$
$\beta = 101.74 \ (1)^{\circ}$	T = 293 (2) K
$\gamma = 100.64 \ (1)^{\circ}$	Prism, pale red
$V = 863.3 (2) \text{ Å}^3$	$0.3 \times 0.1 \times 0.1 \text{ mm}$

Data collection

Enraf-Nonius TurboCAD-4 $R_{\rm int} = 0.016$ $\theta_{\rm max} = 25.0^{\circ}$ diffractometer $h = 0 \rightarrow 6$ Non-profiled $\omega/2\theta$ scans Absorption correction: ψ scan $k = -13 \rightarrow 13$ (North et al., 1968) $l = -16 \rightarrow 15$ $T_{\min} = 0.856, T_{\max} = 0.895$ 3 standard reflections 3374 measured reflections frequency: 166 min 3036 independent reflections intensity decay: none 2657 reflections with $I > 2\sigma(I)$

Refinement

Refinement on F^2 $w = 1/[\sigma^2(F_0^2) + (0.0257P)^2]$ $R[F^2 > 2\sigma(F^2)] = 0.026$ + 0.59P $wR(F^2) = 0.063$ where $P = (F_0^2 + 2F_c^2)/3$ S = 1.04 $(\Delta/\sigma)_{\rm max} = 0.004$ $\Delta \rho_{\rm max} = 0.62 \ {\rm e} \ {\rm \AA}^{-3}$ 3036 reflections $\Delta \rho_{\rm min} = -0.47 \ {\rm e} \ {\rm \AA}^{-3}$ 274 parameters All H-atom parameters refined

H atoms were freely refined, resulting in C-H distances of 0.80 (4)–0.99 (3) Å and $U_{iso}(H)$ values of 0.045 (9)–0.080 (14) Å².

Data collection: CAD-4 EXPRESS (Enraf-Nonius, 1994); cell refinement: CAD-4 EXPRESS; data reduction: XCAD4 (Harms & Wocadlo, 1995); program(s) used to solve structure: SIR97 (Altomare et al., 1999); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP3 for Windows (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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References

- Allen, F. H. (2002). Acta Cryst. B58, 380-388.
- Altomare, A., Burla, M. C., Camalli, M., Cascarano, G., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Polidori, G. & Spagna, R. (1999). J. Appl. Cryst. 32, 115–119.
- Belforte, A., Dell'Amico, D. B., Calderazzo, F., Devillers, M. & Englert, U. (1993). Inorg. Chem. 32, 2282–2286.

- Bruno, I. J., Cole, J. C., Edgington, P. R., Kessler, M., Macrae, C. F., McCabe, P., Pearson, J. & Taylor, R. (2002). Acta Cryst. B58, 389–397.
- Cotton, F. A., Barnard, T. S., Daniels, L. M. & Murillo, C. A. (2002). Inorg. Chem. Commun. 5, 527–532.
- Cotton, F. A., Falvello, L. R. & Murillo, C. A. (1983). *Inorg. Chem.* 22, 382–387. Cotton, F. A. & Walton, R. A. (1993). *Multiple Bonds Between Metal Atoms*,
- 2nd ed. Oxford, New York: Oxford University Press. Enraf-Nonius (1994). *CAD-4 EXPRESS*. Enraf-Nonius, Delft, The Netherlands.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
- Harms, K. & Wocadlo, S. (1995). XCAD4. University of Marburg, Germany.
- Liwporncharoenvong, T. & Luck, R. L. (2002). Inorg. Chim. Acta, 329, 51-58.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). Acta Cryst. A24, 351–359.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.